

OXYGEN, GAS

Material Safety Data Sheet

1. PRODUCT AND COMPANY IDENTIFICATION

Product Name OXYGEN, GAS

Product Code(s) G-1, 1024

UN-No UN1072

Recommended Use Compressed gas.

Synonyms LASER Oxygen; Oxygen, Compressed

Supplier Address Linde LLC

575 Mountain Ave. Murray Hill, NJ 07974 Phone: 908-464-8100 www.lindeus.com

Linde Gas Puerto Rico, Inc. Las Palmas Village

Road No. 869, Street No. 7 Catano, Puerto Rico 00962 Phone: 787-641-7445 www.pr.lindegas.com

Linde Canada Limited 5860 Chedworth Way Mississauga, Ontario L5R 0A2 Phone: 905-501-1700 www.lindecanada.com

For additional product information contact your local customer service.

Chemical Emergency Phone

Number

Chemtrec: 1-800-424-9300 for US/703-527-3887 outside US

2. HAZARDS IDENTIFICATION

WARNING!

Emergency Overview

Oxidizer

Contact with combustible material may cause fire

Contents under pressure

Keep at temperatures below 52°C / 125°F

Appearance ColorlessPhysical State Compressed gas.Odor Odorless

OSHA Regulatory Status This material is considered hazardous by the OSHA Hazard Communication Standard (29 CFR

1910.1200).

Potential Health Effects

Principle Routes of Exposure Inhalation.

Acute Toxicity

Inhalation Oxygen is not acutely toxic under normal pressure. Oxygen is more toxic when inhaled at elevated

pressures. Depending upon pressure and duration of exposure, pure oxygen at elevated pressures may

cause cramps, dizziness, difficulty breathing, convulsions, edema and death.

Eyes None known. Contact with rapidly expanding gas near the point of release may cause frostbite.

Skin None known. Contact with rapidly expanding gas near the point of release may cause frostbite.

Skin Absorption Hazard No known hazard in contact with skin.

Ingestion None known.

Chronic Effects Prolonged inhalation of high oxygen concentrations (>75%) may affect coordination, attention, and

cause tiredness of respiratory irritation.

Aggravated Medical

Conditions

Chronic obstructive pulmonary disease.

Environmental Hazard See Section 12 for additional Ecological Information.

3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical Name	CAS-No	Volume %	Chemical Formula
Oxygen	7782-44-7	>99	02

4. FIRST AID MEASURES

Eye Contact None under normal use. Get medical attention if symptoms occur.

Skin Contact None under normal use. Get medical attention if symptoms occur.

Inhalation Move victim to fresh air. Seek immediate medical attention/advice.

Ingestion None under normal use. Get medical attention if symptoms occur.

Notes to Physician Treat symptomatically.

5. FIRE-FIGHTING MEASURES

Flammable Properties Oxidizer. May vigorously accelerate combustion.

Suitable Extinguishing MediaUse extinguishing measures that are appropriate to local circumstances and the surrounding

environment.

Explosion Data

Sensitivity to Mechanical Impact None.

Sensitivity to Static Discharge None.

Specific Hazards Arising from the

Chemical

May ignite combustibles (wood paper, oil, clothing, etc.) High oxygen concentrations vigorously accelerate combustion. Cylinders may rupture under extreme heat. Continue to cool fire exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

Protective Equipment and Precautions for Firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions Ensure adequate ventilation. Monitor oxygen level.

Environmental Precautions Prevent spreading of vapors through sewers, ventilation systems and confined areas.

Methods for Containment Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. If leak is in

container or container valve, contact the appropriate emergency telephone number in Section 1 or call

your closest Linde location.

Methods for Cleaning Up Return cylinder to Linde or an authorized distributor.

7. HANDLING AND STORAGE

Handling

Dry product is non-corrosive and may be used with all materials of construction. Moisture causes metal oxides which are formed with air to be hydrated so that they include volume and lose their protective role (rust formation). Concentrations of SO₂, Cl₂, salt, etc. in the moisture enhances the rusting of metals in air. Carbon steels and low alloy steels are acceptable for use at lower pressures. For high pressure applications stainless steels are acceptable as are copper and its alloys, nickel and its alloys, brass bronze, silicon alloys, Monel®, Inconel®, and beryllium. Lead and silver or lead tin alloys are good gasket materials. Teflon®, Teflon® composites, or Kel-F® are preferred non-metallic gasket materials.

Oxygen should not be used as a substitute for compressed air in pneumatic equipment since they generally conatin flammable lubricants. Equipment able to use oxygen must be "cleaned for oxygen service". Check with the equipment supplier to verify oxygen compatibility for the service conditions.

Stationary customer site vessels should be operated in accordance with the manufacturer's and Linde's instruction. Do not attempt to repair, adjust or in any other way modify the operation of these vessels. If there is a malfunction or other type of operations problem with the vessel, contact the closest Linde location immediately for assistance. "NO SMOKING" signs should be posted in storage and use areas. Containers of liquid oxygen should be separated from flammable gas containers by a minimum distance of 20 ft., or by a barrier of non-combustible material at least 5 ft. high having a fire resistance rating of 1/2 hour.

Use only in ventilated areas. Never attempt to lift a cylinder by its valve protection cap. Protect cylinders from physical damage; do not drag, roll, slide or drop. When moving cylinders, even for short distance, use a cart designed to transport cylinders. Use equipment rated for cylinder pressure. Use backflow preventive device in piping. Never insert an object (e.g. wrench, screwdriver, pry bar,etc.) into valve cap openings. Doing so may damage valve, causing leak to occur.

Use an adjustable strap wrench to remove over-tight or rusted caps. Close valve after each use and when empty. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier.

Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never attempt to refill a compressed gas cylinder without the owner's written consent. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit.

For additional recommendations, consult Compressed Gas Association's Pamphlets SB-7, G-4.3, G-4.1, G-4.4, P-2.5, G-4.9, P-14, and SB-2.

Storage Protect from physical damage. Cylinders should be stored upright with valve protection cap in place

and firmly secured to prevent falling. Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Keep at temperatures below 52°C / 125°F. Full and empty cylinders should be segregrated. Use a "first in-first out" inventory system to prevent full cylinders from being stored or excessive periods of time. Always store and handle compressed gas cylinders in accordance with Compressed Gas Association, pamphlet CGA-P1,

Safe Handling of Compressed Gases in Containers.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Exposure GuidelinesThis product does not contain any hazardous materials with occupational exposure limits established

by the region specific regulatory bodies.

Engineering Measures Showers. Eyewash stations. Ventilation systems.

VentilationUse local exhaust in combination with general ventilation as necessary to keep oxygen concentrations

below 23.5%.

Personal Protective Equipment

Eye/Face Protection Wear protective eyewear (safety glasses).

Skin and Body Protection Work gloves and safety shoes are recommended when handling cylinders. Gloves must be clean and

free from grease or oil.

Respiratory Protection

General Use No special protective equipment required.

Hygiene Measures Handle in accordance with good industrial hygiene and safety practice.

9. PHYSICAL AND CHEMICAL PROPERTIES

AppearanceColorless.OdorOdorless.Odor ThresholdNo information availablePhysical StateCompressed gasFlash PointNo information availableAutoignition TemperatureNo information available

Flash PointNo information available.Autoignition TemperatureNo information availableDecomposition TemperatureNo information availableBoiling Point/Range-182.9°C / -297.3°FFreezing Point-218.8°C / -361.8°FMolecular Weight32.00

Water SolubilitySlightly solubleEvaporation RateNo information availableVapor PressureAbove critical temp.Vapor Density1.326 kg/m³ (0.083 lb/ft³)

@21.1°C

VOC Content(%) Not applicable **Partition Coefficient: n-** Log P -0.65

octanol/water

Specific Vol. @ 21.1°C & 1 atm 12.1 ft³/lb Critical Pressure 731.4 psia

Critical Temperature -118.57°C / -215.4°F Flammability Limits in Air

UpperNot applicableLowerNot applicable

10. STABILITY AND REACTIVITY

Stability Stable.

Incompatible Products Combustible materials. Organic material. Reducing agents.

Conditions to Avoid Keep away from open flames, hot surfaces and sources of ignition.

Hazardous Decomposition

Products

None known.

Hazardous Polymerization Hazardous polymerization does not occur.

11. TOXICOLOGICAL INFORMATION

Acute Toxicity

LD50 Oral: No information available.

LD50 Dermal: No information available.

LC50 Inhalation: No information available.

Inhalation Human volunteers which inhaled 90-95% oxygen through a face mask for 6 hours showed signs of

tracheal irritation and fatigue. Other symptoms (which may have been caused by placing a tube into the trachea during the experiment) included: sinusitis, conjunctivitis, fever, and symptoms of acute

bronchitis.

Poisoning began in dogs 36 hours after inhalation of pure oxygen at atmospheric pressure. Distress

was seen within 48 hours and death within 60 hours.

Eye Contact The incompletely developed retinal circulation is more susceptible to toxic levels of oxygen. In

premature infants, arterial oxygen tension above 150 mm Hg may cause retrolental fibroplasia. Permanent blindness may occur several months later. One case of severe retinal damage in an adult was reported. An individual suffering from myasthenia gravis developed irreversible retinal atrophy

after breathing 80% oxygen for 150 days.

Repeated Dose ToxicityNo information available.

Chronic Toxicity

Chronic Toxicity Prolonged inhalation of high oxygen concentrations (>75%) may affect coordination, attention, and

cause tiredness of respiratory irritation.

Carcinogenicity Contains no ingredient listed as a carcinogen.

Irritation No information available.

Sensitization No information available.

Reproductive Toxicity No information available.

Developmental ToxicityNo information available.

Synergistic Materials None known.

Target Organ Effects None known.

12. ECOLOGICAL INFORMATION

Ecotoxicity

Will not bioconcentrate.

Ozone depletion potential; ODP; (R-11 = 1): Does not contain ozone depleting chemical (40 CFR Part 82).

13. DISPOSAL CONSIDERATIONS

Waste Disposal Methods Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container

PROPERLY LABELED WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP IN

PLACE to Linde for proper disposal.

14. TRANSPORT INFORMATION

DOT

Proper Shipping Name Oxygen, compressed

Hazard Class2.2Subsidiary Class5.1UN-NoUN1072

Description UN1072,0xygen, compressed,2.2,(5.1)

TDG

Proper Shipping Name Oxygen, compressed

Hazard Class 2.2 Subsidiary Class (5.1) UN-No UN1072

Description UN1072,OXYGEN, COMPRESSED,2.2(5.1)

MEX

Proper Shipping Name Oxygen, compressed

Hazard Class2.2Subsidiary Class5.1UN-NoUN1072

Description UN1072 Oxygen, compressed,2.2

IATA

UN-No UN1072

Proper Shipping Name Oxygen, compressed

Hazard Class 2.2 Subsidiary Class 5.1 ERG Code 2X

Description UN1072,Oxygen, compressed,2.2(5.1)

Maximum Quantity for Passenger75 kgMaximum Quantity for Cargo Only150 kg

Limited Quantity

No information available.

IMDG/IMO

Proper Shipping Name Oxygen, compressed

Hazard Class 2.2

Subsidiary Class5.1UN-NoUN1072EmS No.F-C, S-W

Description UN1072, Oxygen, compressed, 2.2(5.1)

ADR

Proper Shipping Name Oxygen, compressed

Hazard Class2.2UN-NoUN1072Classification Code10

Description UN1072 Oxygen, compressed, 2.2,

ADR/RID-Labels 5.1

15. REGULATORY INFORMATION

International Inventories

TSCA Complies
DSL Complies
EINECS/ELINCS Complies

Legend

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory

DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

EINECS/ELINCS - European Inventory of Existing Commercial Chemical Substances/EU List of Notified Chemical Substances

U.S. Federal Regulations

SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372.

SARA 311/312 Hazard Categories

Acute Health HazardNoChronic Health HazardNoFire HazardYesSudden Release of Pressure HazardYesReactive HazardNo

Clean Water Act

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42).

Risk and Process Safety Management Programs

This material, as supplied, does not contain any regulated substances with specified thresholds under 40 CFR Part 68. This product does not contain any substances regulated as Highly Hazardous Chemicals pursuant to the 29 CFR Part 1910.110.

Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61)

This product does not contain any substances regulated as hazardous air pollutants (HAPS) under Section 112 of the Clean Air Act Amendments of 1990.

CERCLA/SARA

This material, as supplied, does not contain any substances regulated as hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) or the Superfund Amendments and Reauthorization Act (SARA) (40 CFR 355). There may be specific reporting requirements at the local, regional, or state level pertaining to releases of this material.

U.S. State Regulations

California Proposition 65

This product does not contain any Proposition 65 chemicals.

U.S. State Right-to-Know Regulations

Chemical Name	Massachusetts	New Jersey	Pennsylvania	Illinois	Rhode Island
Oxygen	Χ	Χ	Χ		Χ

International Regulations

Canada

This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all the information required by the CPR.

WHMIS Hazard Class

A Compressed gases C Oxidizing materials



16. OTHER INFORMATION

Prepared By Product Stewardship

23 British American Blvd. Latham, NY 12110 1-800-572-6501

Issuing Date 05-Mar-2010

Revision Date

Revision Number 0

Revision Note Initial Release.

NFPA Health Hazard 0 Flammability 0 Stability 0 Physical and Chemical

Hazards OX

HMIS Health Hazard 0 Flammability 0 Physical Hazard 3 Personal Protection -

Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2009, CGA Recommended Hazard Ratings for Compressed Gases, 3rd Edition.

General Disclaimer

For terms and conditions, including limitation of liability, please refer to the purchase agreement in effect between Linde LLC, Linde Merchant Production, Inc. or Linde Gas North America LLC (or any of their affiliates and subsidiaries) and the purchaser.

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End of Safety Data Sheet